EVIDENCE FOR THE FORMATION OF A X-REARRANGED

ALLYLIC PHOSPHONIUM SALT

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Although the conversion of α -halogenoketones to phosphonium salts is well documented¹, nothing is known about similar transformations in γ -halogeno- $\alpha_i \beta$ -unsaturated ketones, e.g. (<u>E</u>)-5-bromo-3-penten-2-one, <u>1</u>.

Enriched crudes of <u>1</u> react satisfactorily with Ph_3P . These crudes can be obtained either by allylic bromination of pure (preparative g.l.c.) (<u>E</u>)-3-penten-2-one² or by Wittig condensation of bromoacetaldehyde³ and ace-tylmethylenetriphenylphosphorane. A mixture of two isomeric phosphonium salts (correct elemental analysis) is produced in the reaction of <u>1</u> with Ph_3P , as seen by n.m.r.:

 $\begin{array}{c} \text{Ph}_{3} \text{P}_{2}\text{CH}_{2}\text{CH}_{2}\text{CH}_{2}\text{CH}_{2}\text{CH}_{2}\text{CH}_{2}\text{CH}_{2}\text{CH}_{2}\text{CH}_{3} \text{ Br}^{-} \underbrace{2}_{3} \\ \\ \begin{array}{c} \text{Ph}_{3} \text{P}_{2}\text{CH}_{2}\text{CH}_{2}\text{CH}_{2}\text{CH}_{2}\text{CH}_{2}\text{CH}_{2}\text{CH}_{3} \\ \\ \begin{array}{c} \text{CH}_{2} \text{-CH}_{-}\text{CH}_{-}\text{COCH}_{3} \\ \\ \begin{array}{c} \text{Ph}_{3} \text{P}_{3} \end{array} \end{array} \end{array} \right)$

Scheme 1

4-oxo-2-pentenyltriphenylphosphonium bromide, $\underline{2}$, ($\underline{5}$ -2.2, CH₃; 6.8, CH=CH) and 1-vinyl=2-oxopropyltriphenylphosphonium bromide, $\underline{3}$, ($\underline{5}$ -2.3, CH₃; 4.3, -CH₂) in a 3:1 ratio.

Treatment of this mixture of phosphonium salts with t-BuOK and condensation with benzaldehyde afforded a t.l.c. separable mixture of $(3\underline{E},5\underline{E})$ -6-phenyl-3,5-hexadien-2-one⁴, 4, and (\underline{Z}) -3-benzylidene-4-penten-2-one, 5, [6-2.45 (3H,s); 5.4 (2H, J₁₃=18 Hz, J₂₃=11 Hz and J₁₂=2 Hz); 6.5 (1H, J₃₁=18 Hz and J₃₂=11 Hz); 7.3 (1H,s); 7.47 (5H,s)] in the same ratio of the original salts (3:1).

This is the first time that in the reaction between an allyl bromide derivative and Ph_3P both possible phosphonium salts, i.e. "normal" and "rearranged" have been detected.

y-Condensation of an allylic phosphonium bromide has been described by Corey and Erickson⁵ in the condensation of the phosphorane derivative from pure (n.m.r.) (<u>E</u>)-3-methoxy-2-methylallyltriphenylphosphonium bromide with <u>n</u>-hexanal giving methyl 2-isopropenyl-2-octenoate along with the <u>x</u>-condensation products.

One of the three paths proposed by Corey to explain this y-condensation involves allylic rearrangement of Ph₃P group before normal Wittig condensation, although the rearranged phosphonium salt was actually never observed. On the other hand, Castells et al.⁶ had previously described a similar condensation between the phosphorane from 4-bromo-3-methyl-2-butenal and 4-nitrobenzaldehyde assuming that the salt was the rearranged one, since only 3-(4-ni-trophenyl)-2-isopropenyl-2-propenal was isolated. No n.m.r. of the salt could be run. Our finding supports previous rearrangement of the allylic phosphonium bromides. This rearrangement should depend on the electron-withdrawing character of the carbonyl group (CHO>COCH₂>COOCH₂).

Moreover, in the reaction between (\underline{E}) -5-bromo-3-penten-2-one ethyleneketal and Ph₃^P we have always obtained a pure salt, the normal one, (\underline{E}) -4,4-ethylenedioxi-2-pentenyltriphenylphosphonium bromide, <u>6</u>, (correct elemental analysis):

$$\operatorname{BrcH}_{2}^{\mathsf{CH}}\operatorname{E}^{\mathsf{CH}}\operatorname{C}^{\mathsf{C}}\operatorname{CH}_{3} \xrightarrow{\operatorname{Ph}_{3}} \operatorname{Ph}_{3}^{\mathsf{P}}\operatorname{Ph}_{3}^{\mathsf{P}}\operatorname{CH}_{2}^{\mathsf{CH}}\operatorname{E}^{\mathsf{LH}}\operatorname{E}^{\mathsf{C}}\operatorname{C}^{\mathsf{C}}\operatorname{C}_{3}^{\mathsf{H}} \operatorname{Br}^{-} \xrightarrow{\operatorname{H}_{2}^{\mathsf{O}}} \underline{2} \xrightarrow{\operatorname{slowly}} \underline{2} + \underline{3}$$

Hydrolysis of <u>6</u> gives pure <u>2</u>, that in ethereal suspension very slowly converts into a mixture (3:1) of <u>2</u> and <u>3</u>. This fact and the reaction of Scheme 1 suggest two possible mechanisms:

i) a mechanism implying formation of $\underline{2}$ by a S $_N^2$ attack of Ph $_3^P$ and subsequent partial rearrangement of $\underline{2}$ into $\underline{3}$.

ii) the possibility that both salts are formed simultaneously from the collapse of the intermediate Ph_3^{PBr} CH_CH_CHCOCH₃ arising from the nucleophilic attack of Ph_3^{P} to the halogen atom⁷.

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